

SURFACTANTS FLOODING

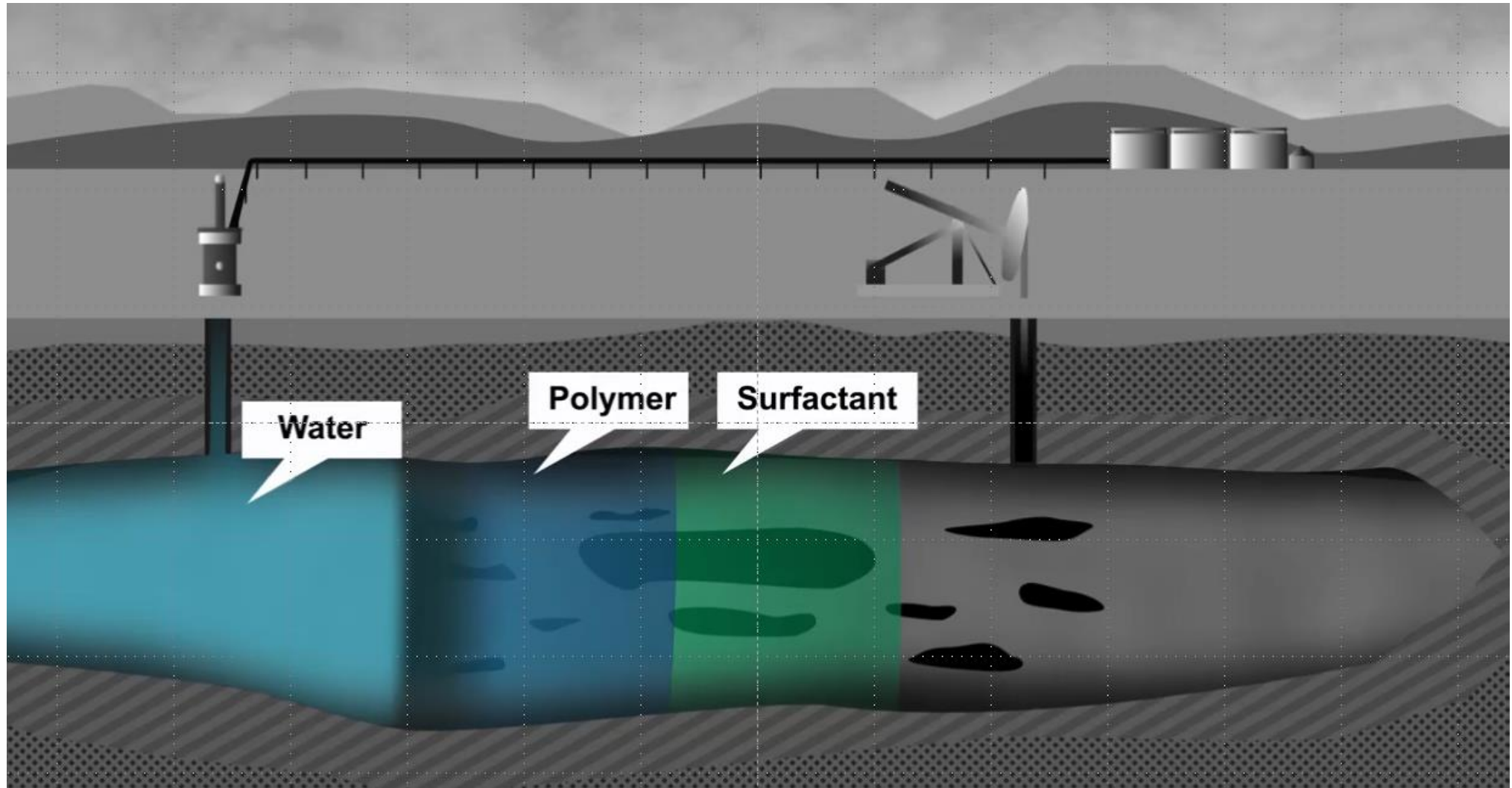
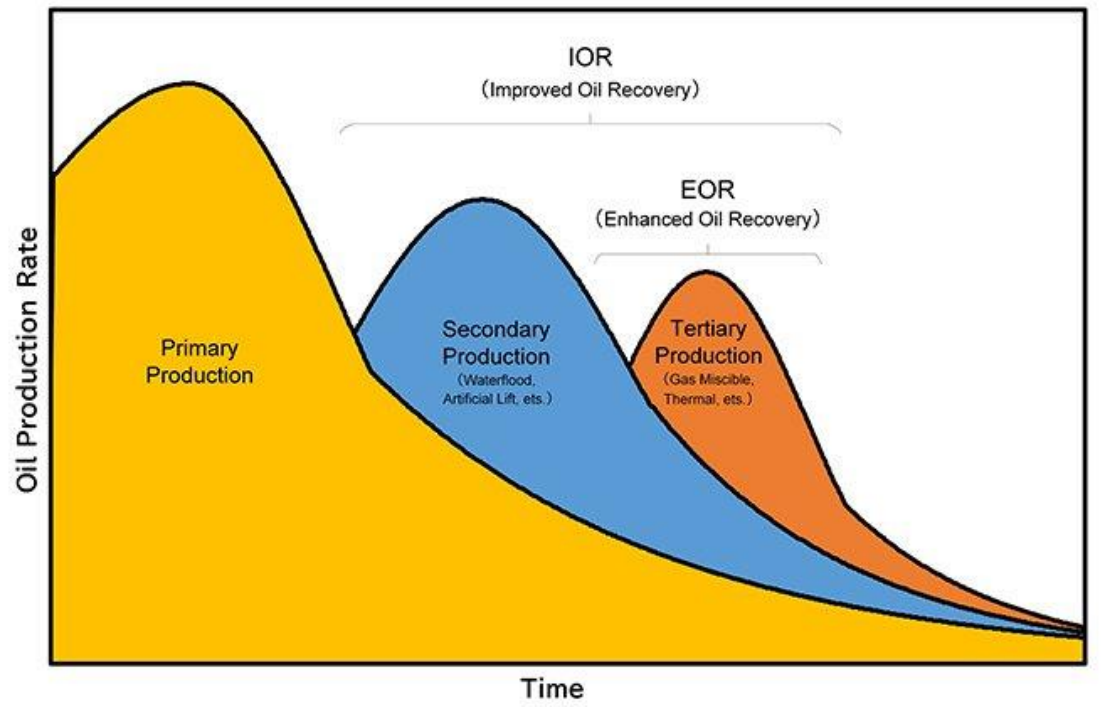
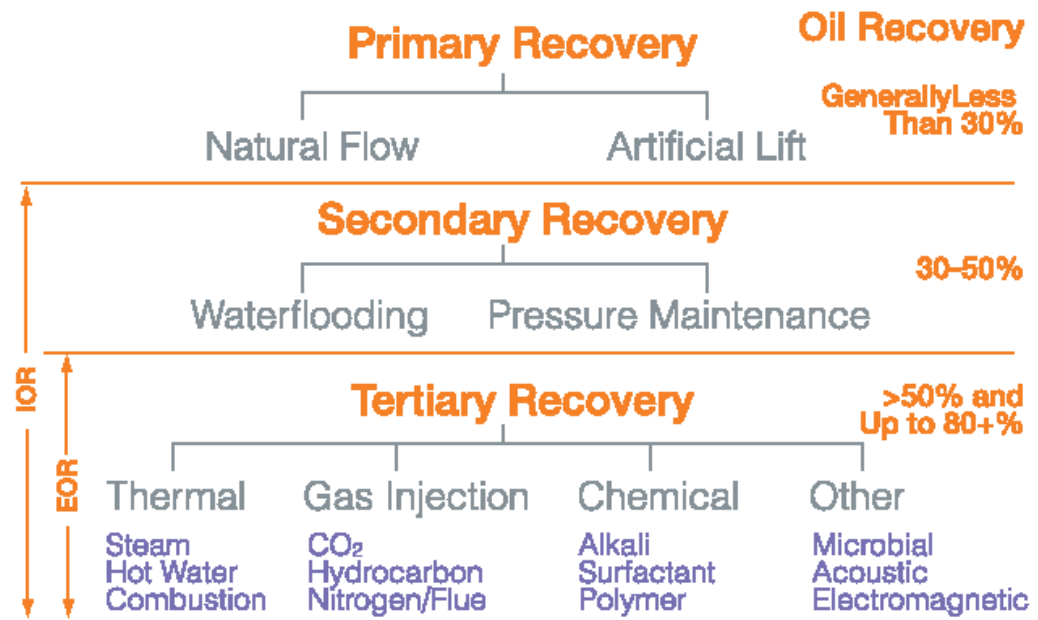
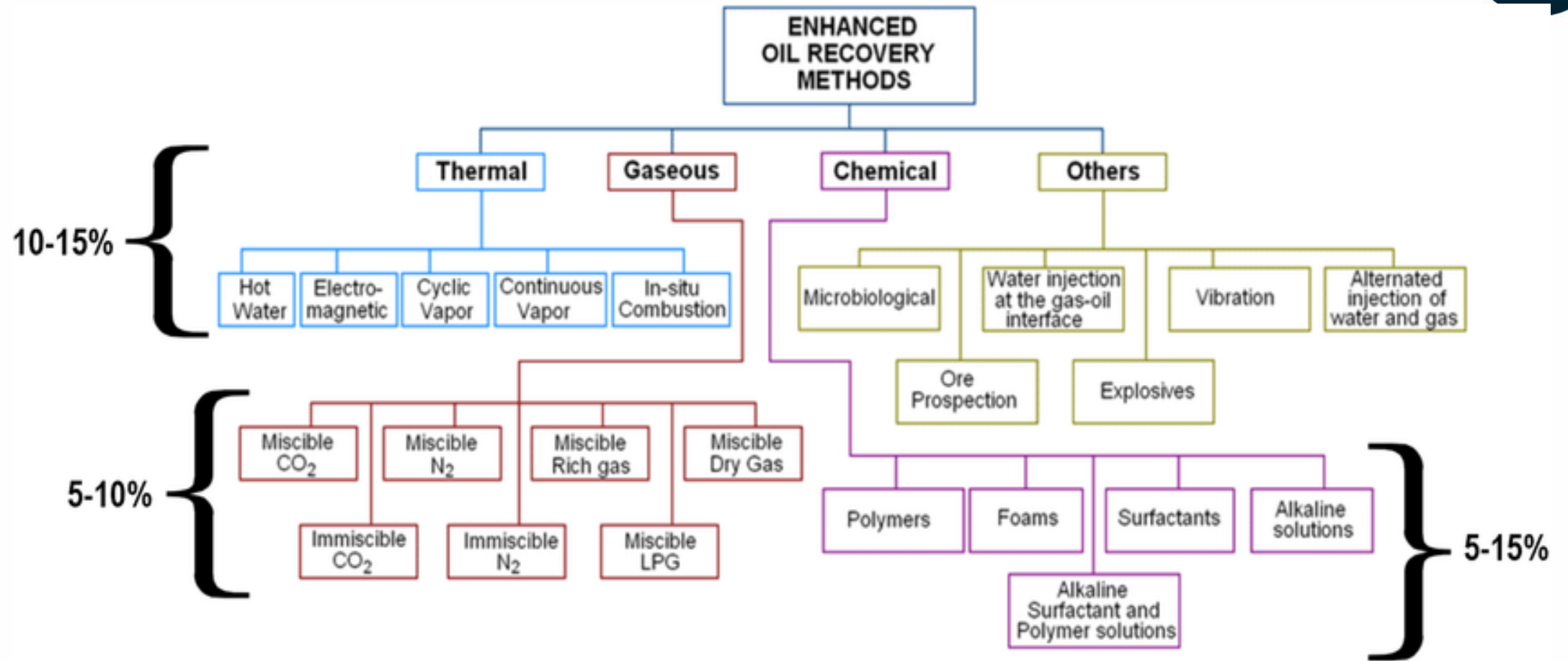


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Oil Recovery





EOR Processes



$$N_c = \frac{F_v}{F_c} = \frac{v\mu}{\sigma \cos \theta}$$

1- Mobility-control

2- Chemical processes

3- Miscible processes

4- Thermal Processes

Screening Criteria for Chemical EOR Processes



1- Formation

2- Oil Composition and Oil Viscosity

3- Formation Water Salinity

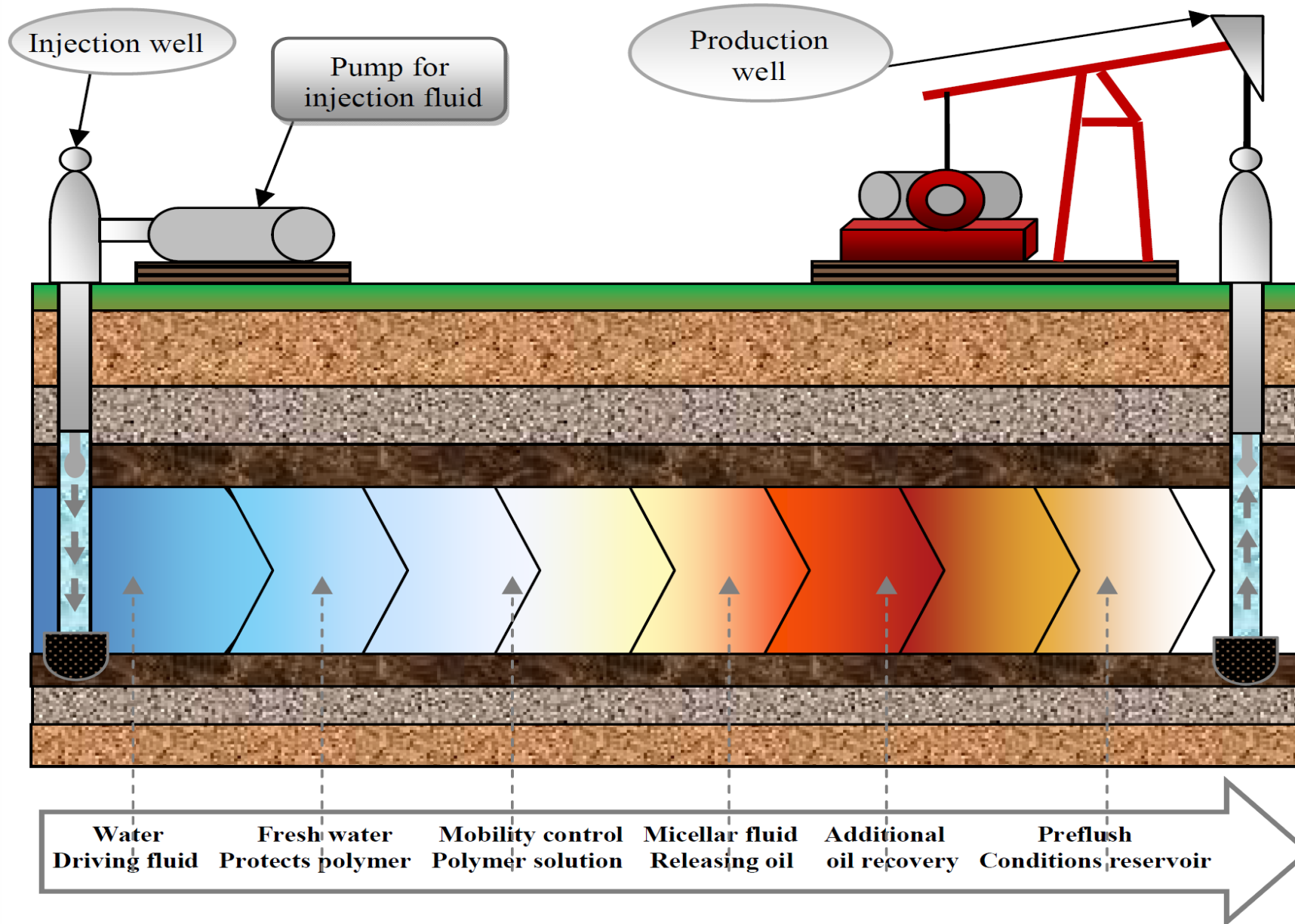
4- Reservoir Temperature

5- Formation Permeability

Mechanism of Surfactant Flooding



- 1- Reducing the IFT between crude oil and brine
- 2- Creating new phase (Microemulsion)
- 3- Change rock wettability



Types of Surfactant Flooding

1- Micelle/polymer flooding

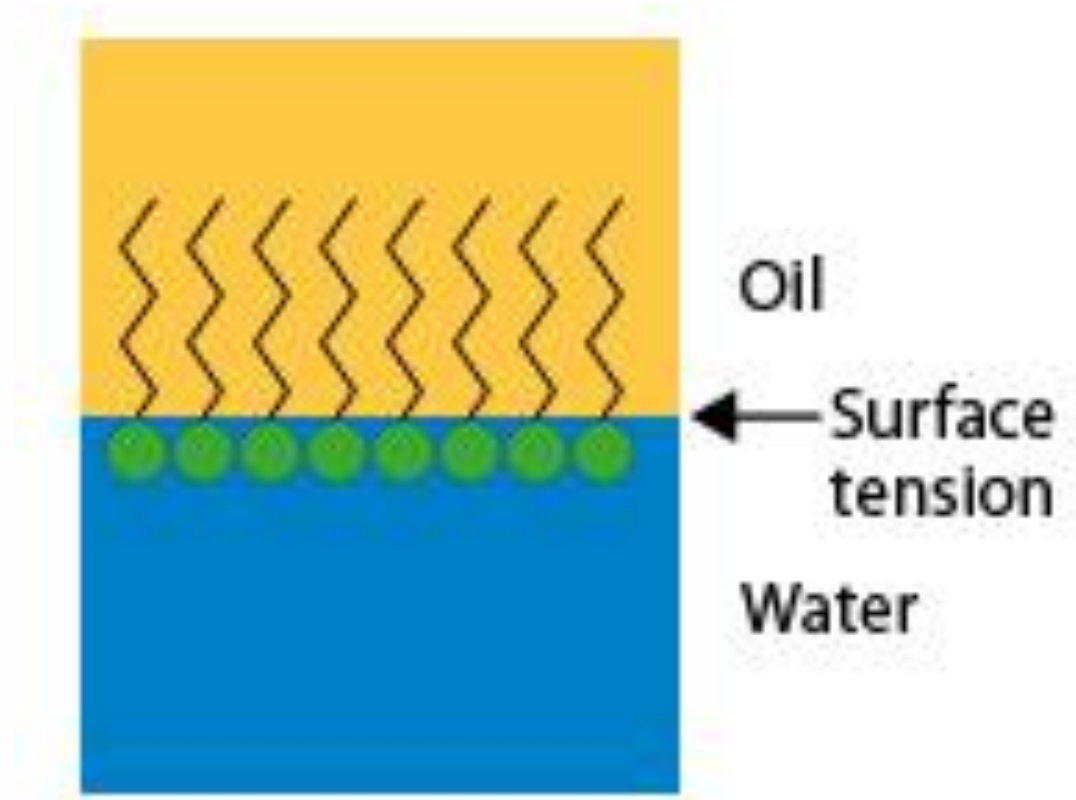
2- Microemulsion flooding

3- Alkaline/surfactant/polymer



SURFACTANTS

Surface Active Agents

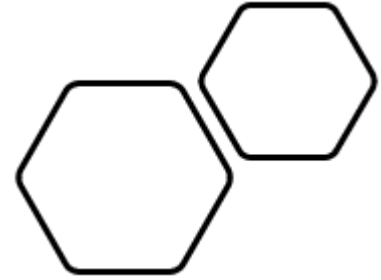


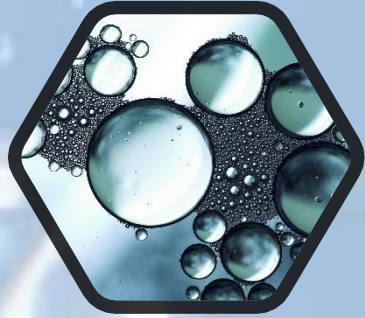


Polar hydrophilic
(water-loving) head



Non-polar hydrophobic
(water-hating) tail





Types

Anionic

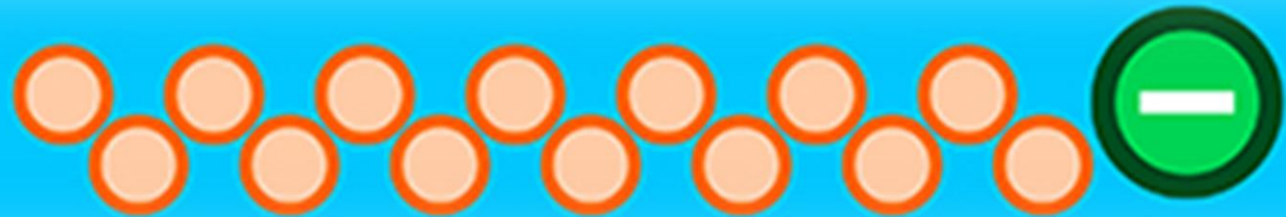
Cationic

Nonionic

zwitterionic

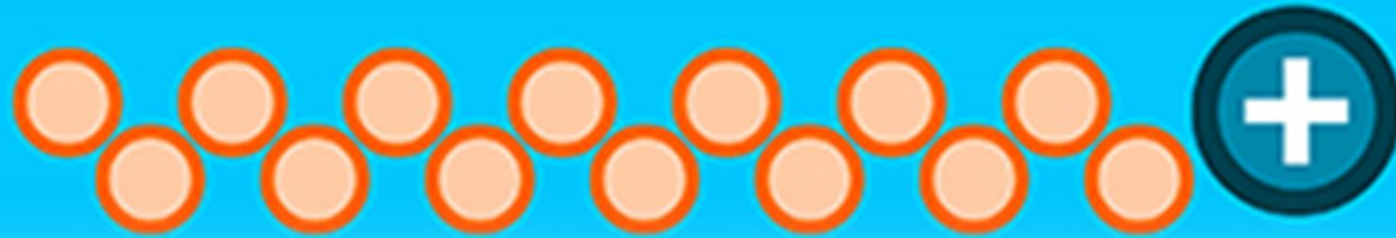


Anionic Surfactant





Cationic Surfactant



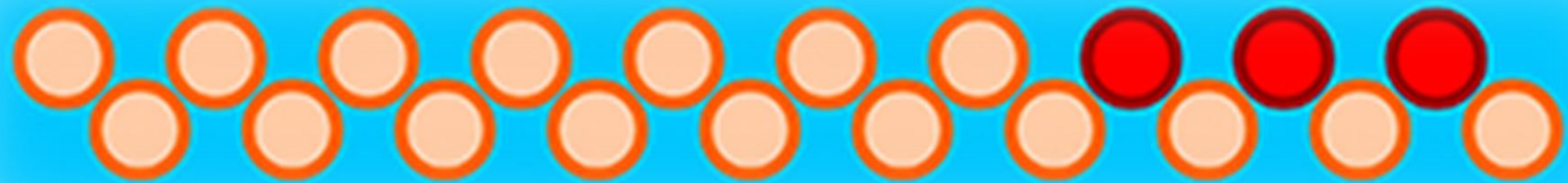


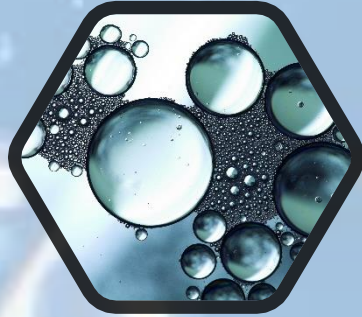
Nonionic Surfactant





Zwitterionic Surfactant





Surfactant Characterization

HLB

CMC

SR

CPP

HL

Hydrophile–Lipophile Balance

hydrophobic

hydrophilic



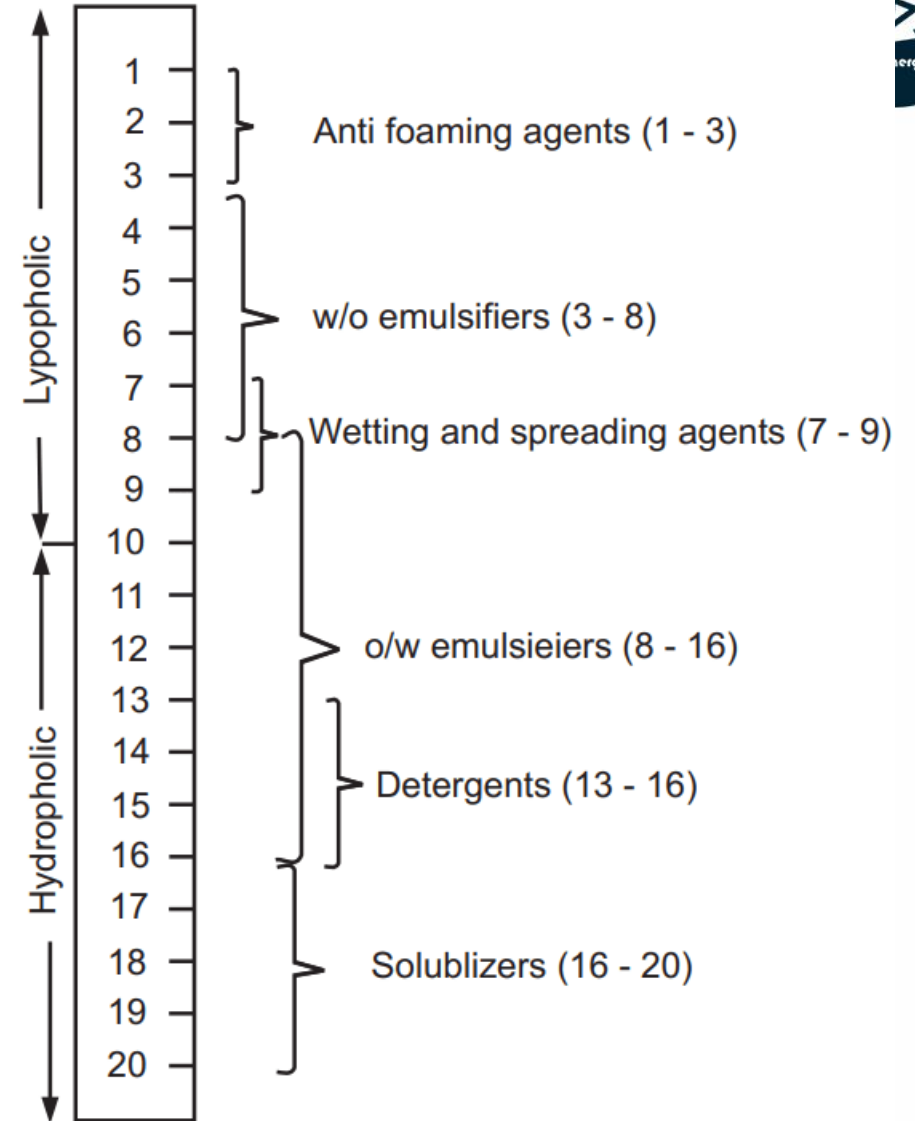


Hydrophile–Lipophile Balance (HLB)

- calculating values for the different regions of the molecule.

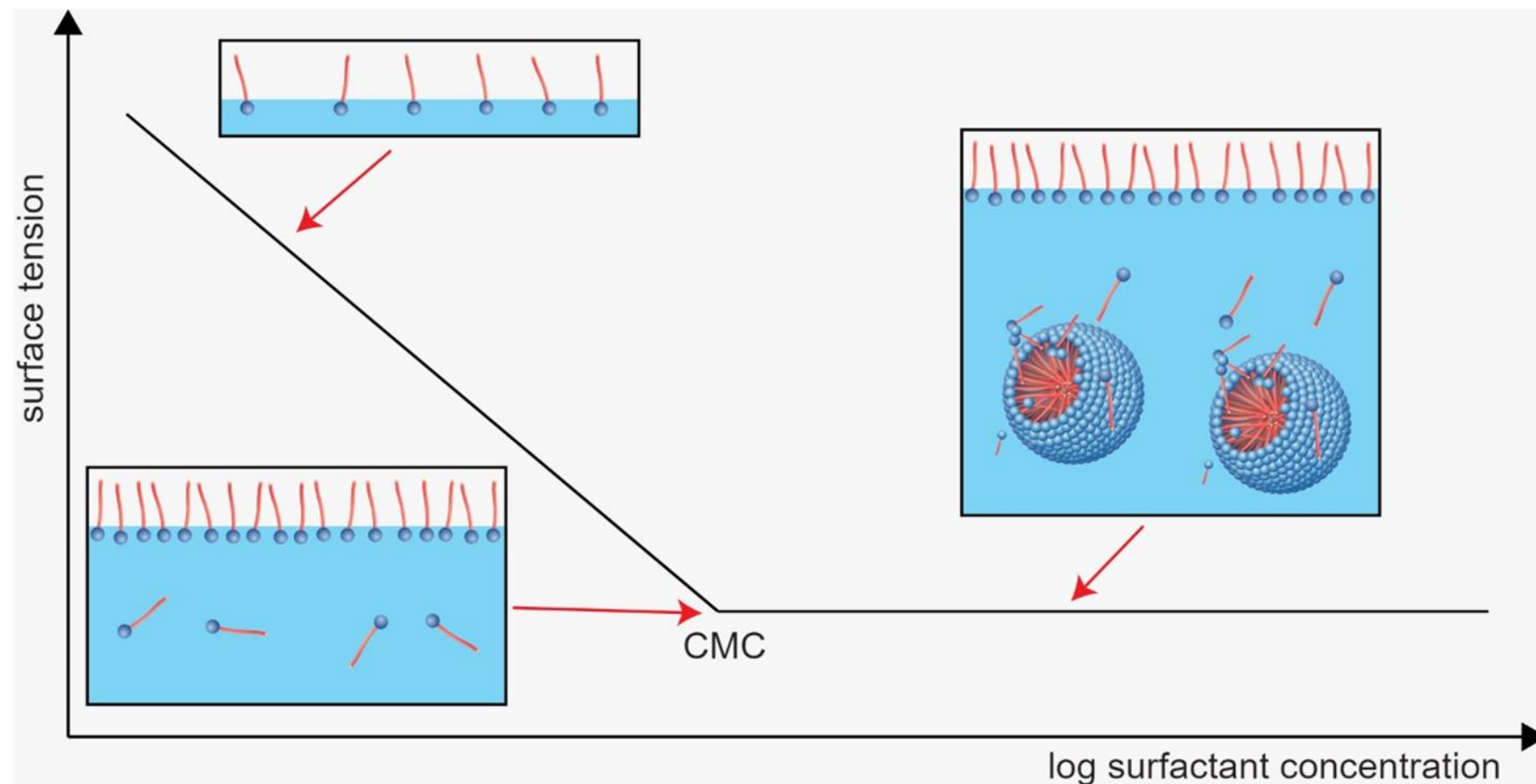
- $$HLB = \frac{20 MW_h}{MW}$$

- MW_h : The molecular mass of the hydrophilic portion of the molecule.
- MW : The molecular mass of the whole molecule



CMC

Critical Micelle Concentration



S

Solubilization Ratio

The ratio of the solubilized oil (water) volume to the surfactant volume in the microemulsion phase.



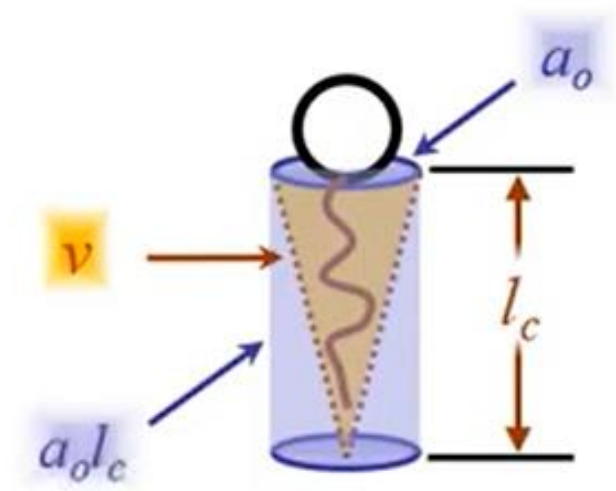
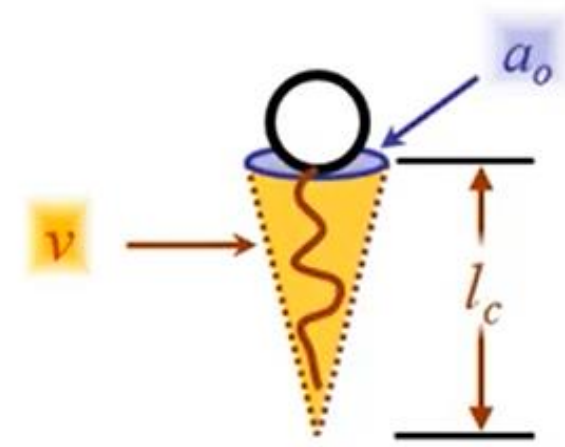
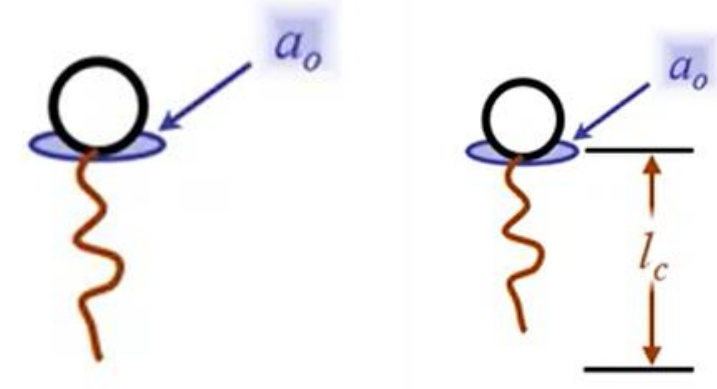
Related to IFT



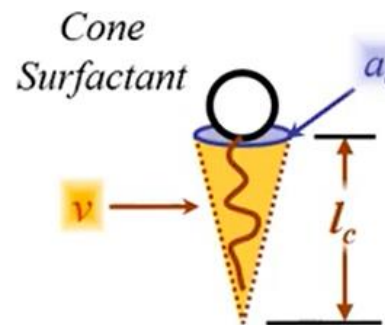
The IFT reaches its minimum

the solubilization ratio for oil is equal to that for water.

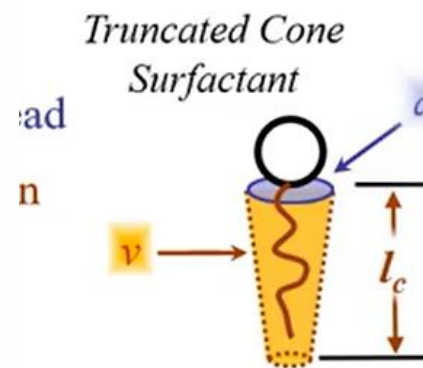
CF Critical Packing Parameter



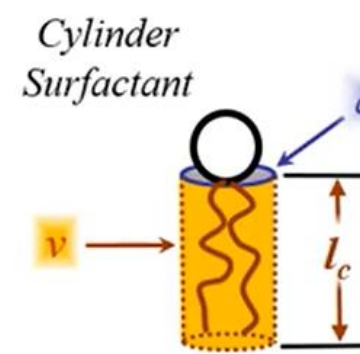
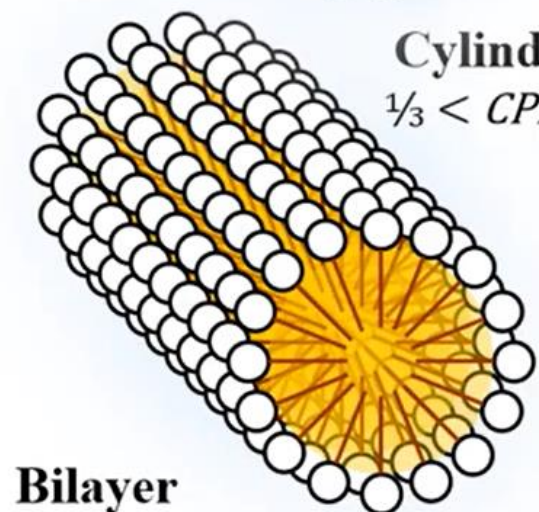
Critical Packing



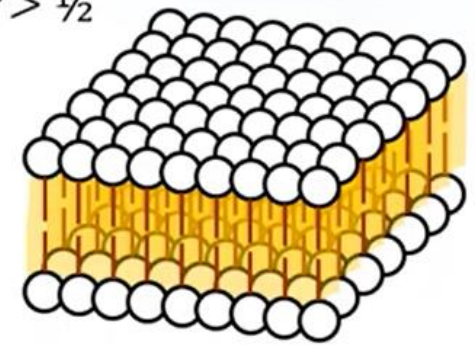
Spherical
 $CPP \leq 1/3$



Cylindrical
 $1/3 < CPP \leq 1/2$



Bilayer
 $CPP > 1/2$



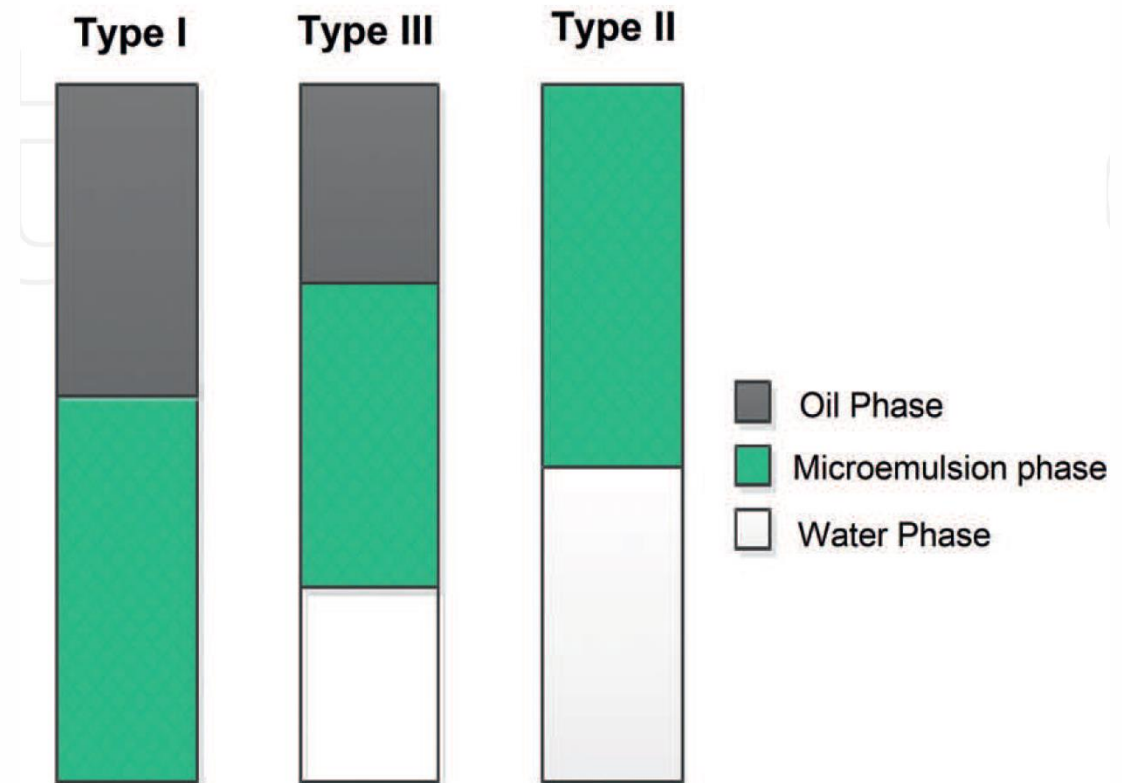
MICROEMULSION

Types of Microemulsion

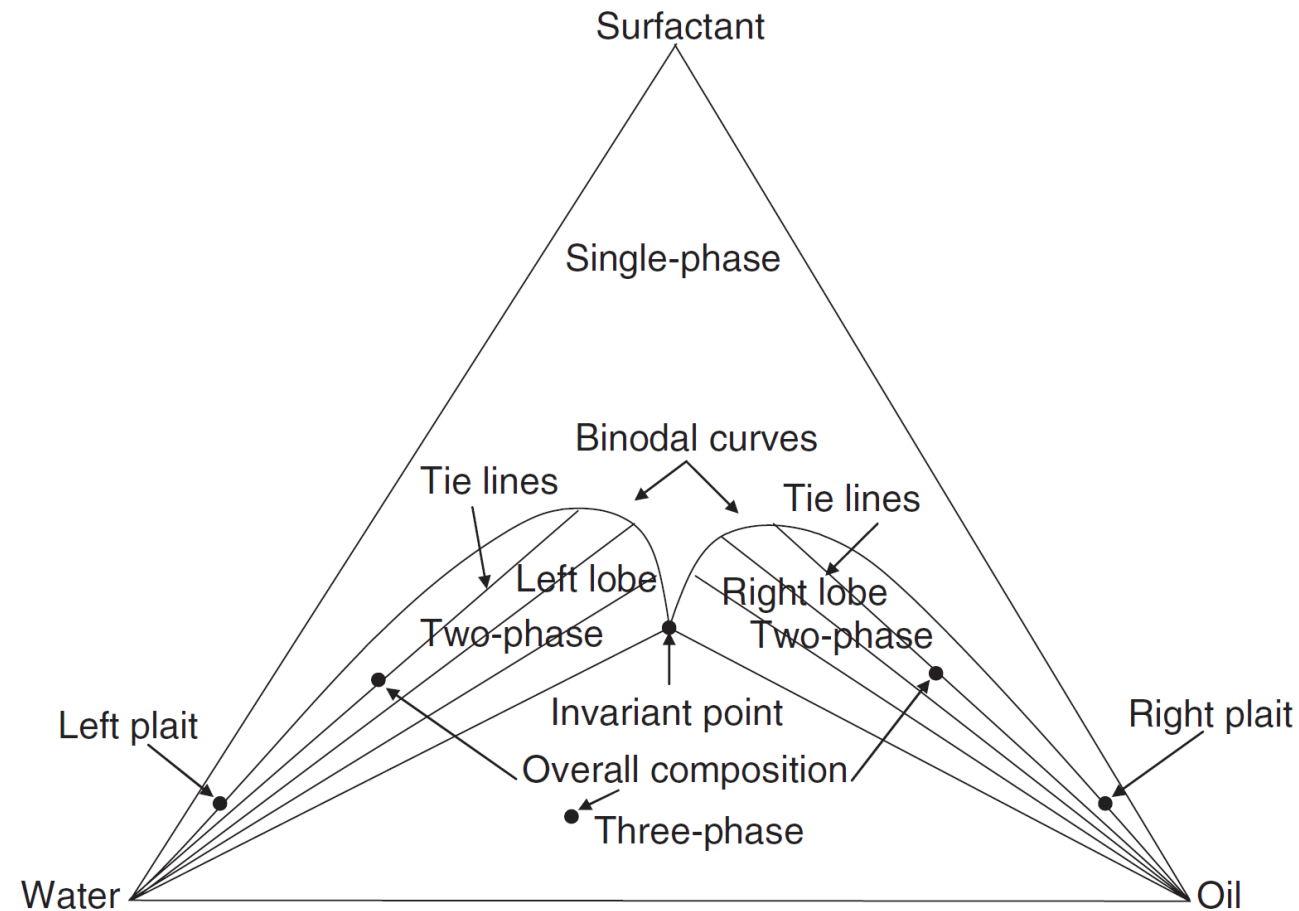
1. Type I (O/W)

2. Type II

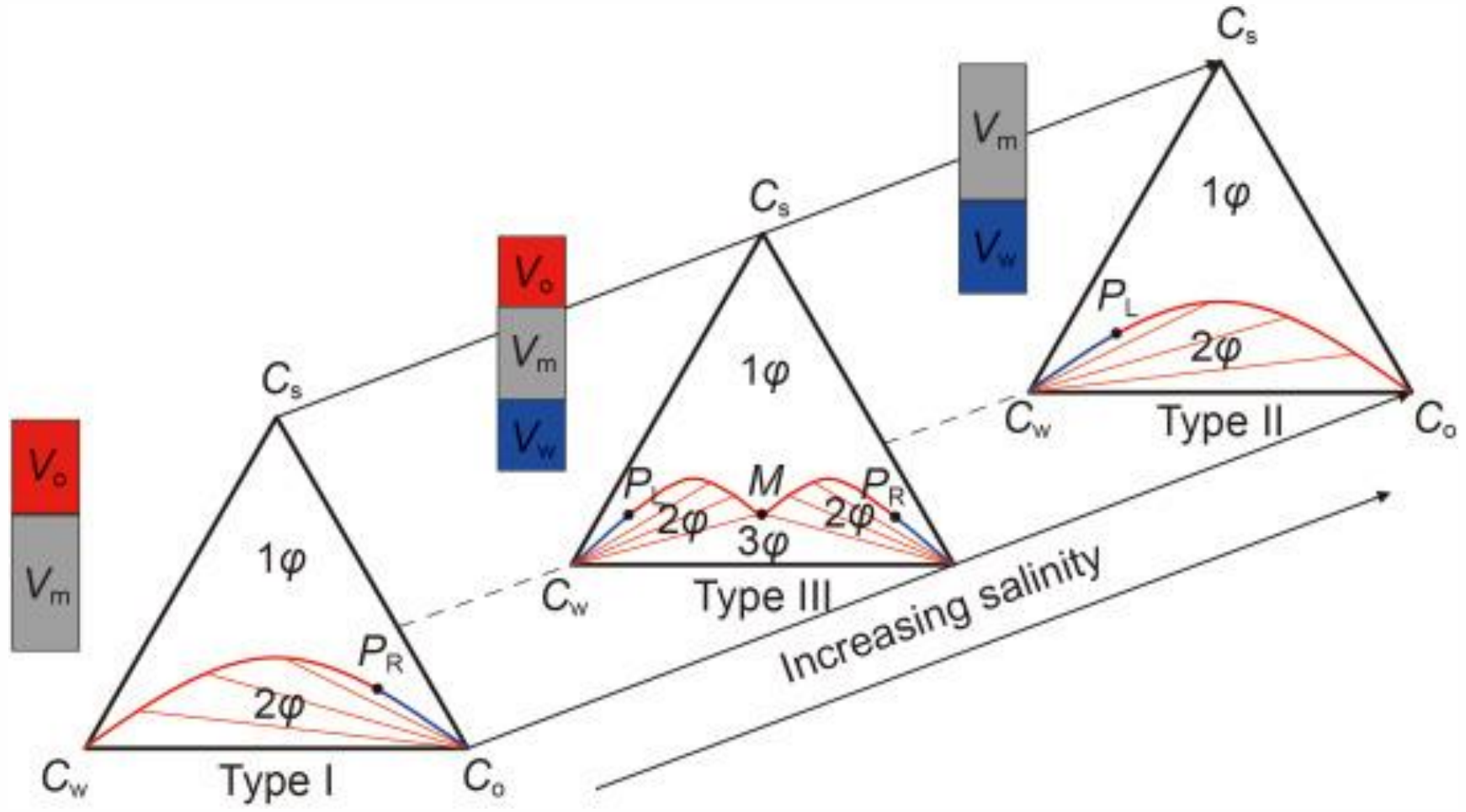
3. Type III



Schematic of a Ternary Diagram

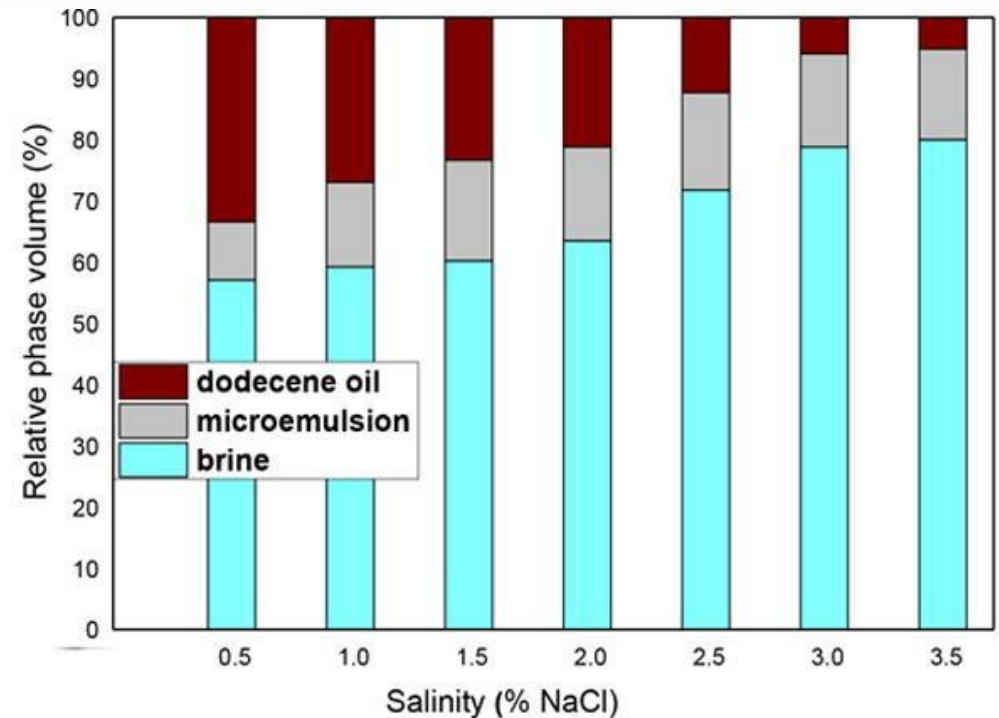
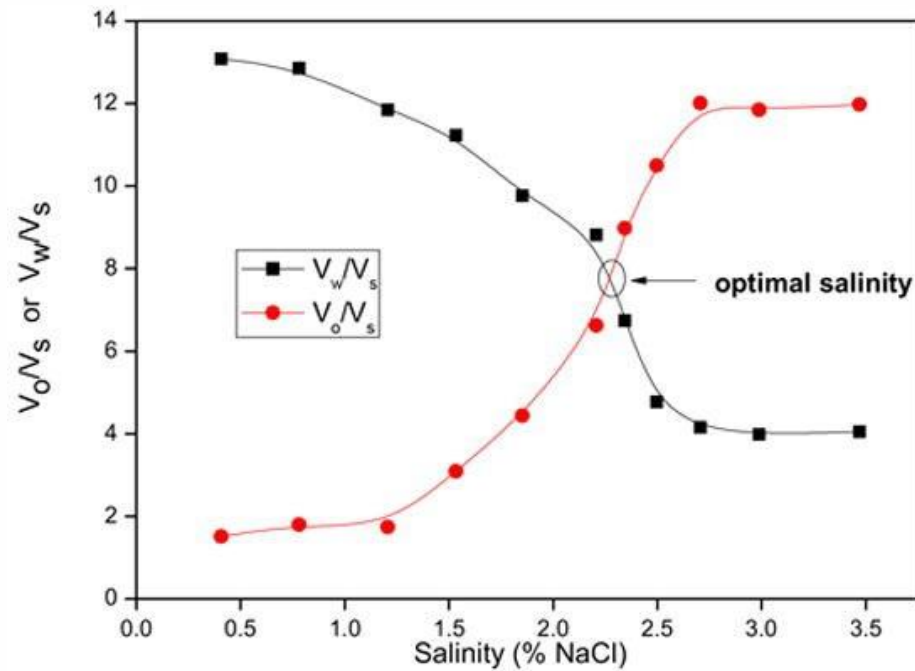


Phase Behavior of Microemulsion

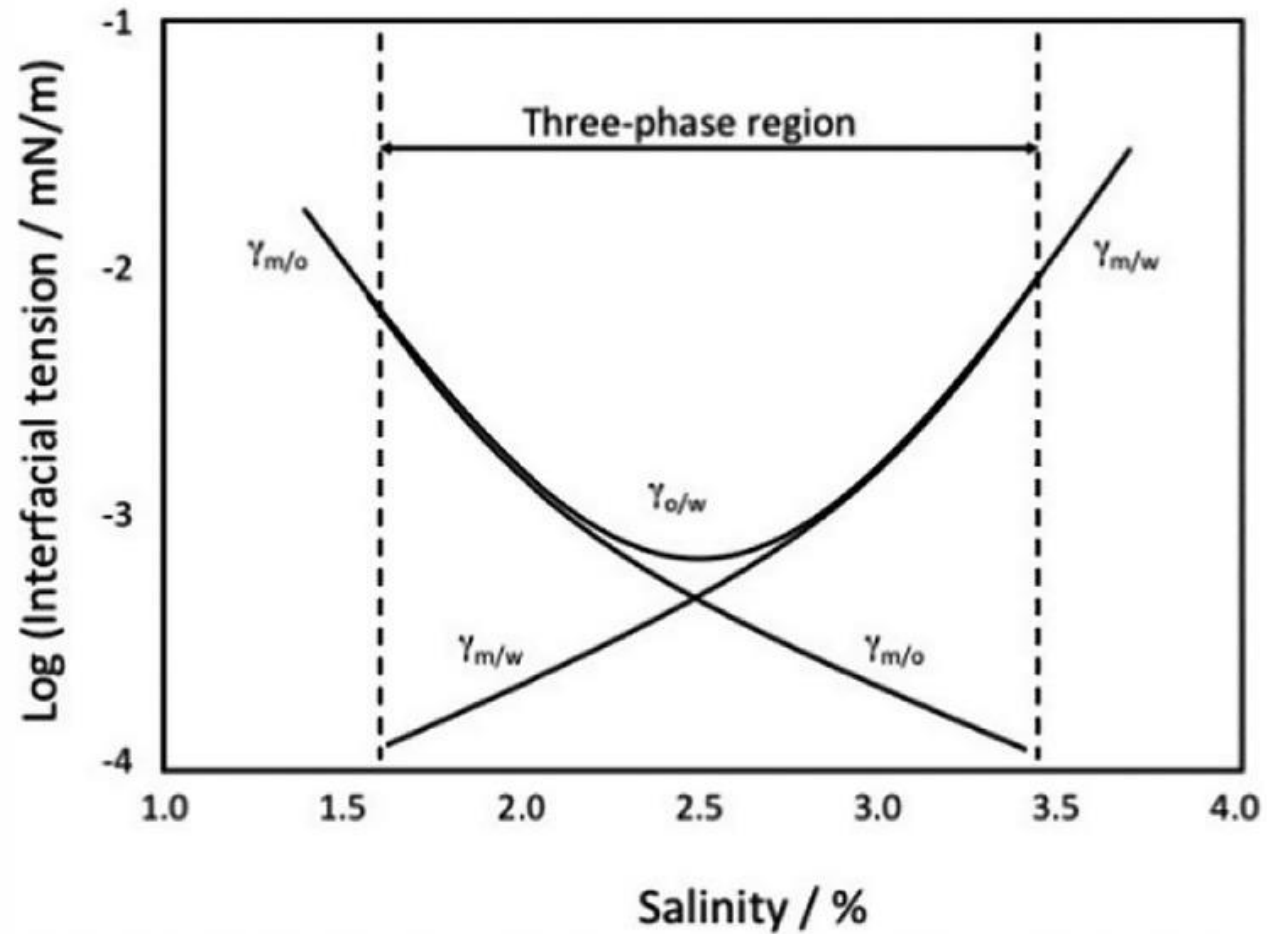


Characteristics of Microemulsion

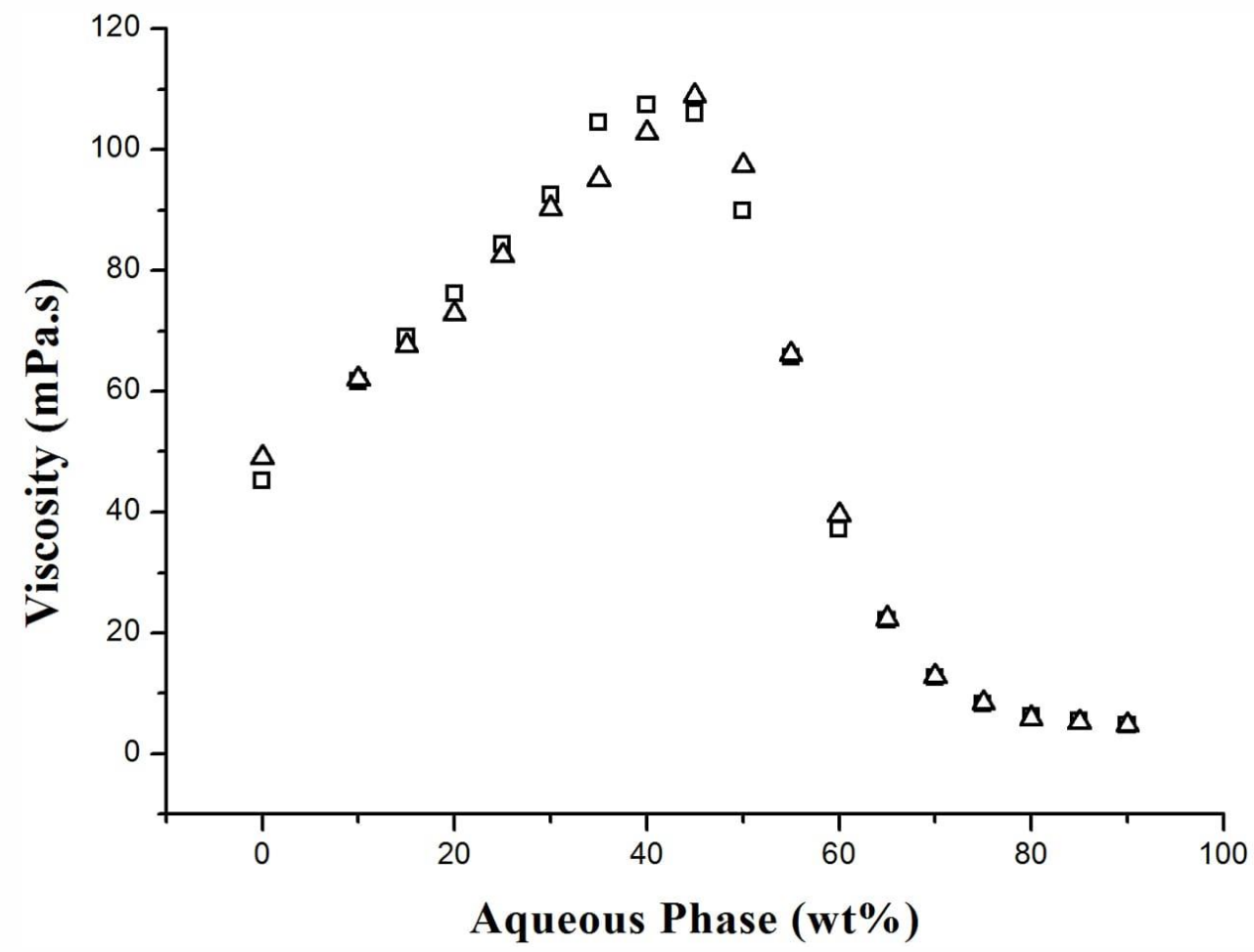
1 Solubilization Capacity



2 IFT Reduction



3 Viscosity and Density



Effect of Chemicals Used

- **Effect of surfactant**
 - Carbon chain length
 - HLB (Hydrophile–Lipophile Balance)

- **Effect of co-surfactant**
 - KM Value

